## **CAMBRIDGE INTERNATIONAL EXAMINATIONS**

**Cambridge International General Certificate of Secondary Education** 

## MARK SCHEME for the October/November 2015 series

## 0653 COMBINED SCIENCE

0653/22

Paper 2 (Core Theory), maximum raw mark 80

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

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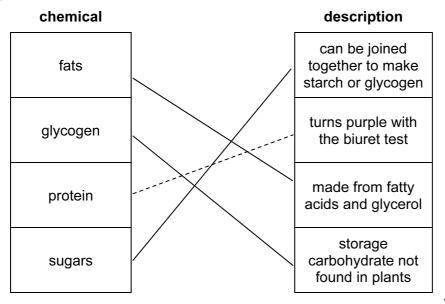
- 1 (a) (i) force X = friction (force); force Y = weight; [2]
  - (ii) 750 N; for constant speed, forces must be balanced/owtte; [2]
  - (b) (i) (30–90 s) constant speed; (90–120 s) (negative) acceleration / deceleration; [2]
    - (ii)  $20 \text{ m/s} = 20 \times 3600 \text{ m/h}$ ; = 72000 m/h = 72 (km/h); [2]

[Total: 8]

- 2 (a) (i) fractional distillation; [1]
  - (ii) because new substances are not made; [1]
  - (b) (i) petroleum fraction boils and vapour moves into delivery tube; vapour condenses in delivery tube; [2]
    - (ii) increasing boiling point from **A** to **D**; [1]
  - (c) (hydrocarbon) + oxygen; water + carbon dioxide; [2]

[Total: 7]

3 (a)



(3 correct for 2 marks, 2 or 1 correct for 1 mark)

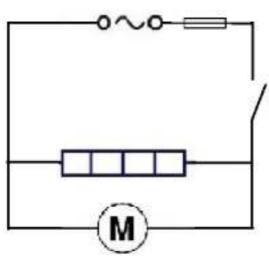
[2]

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- (b) (i) respiration; [1]
  - (ii) (milk A) it contains the highest total of fat, carbohydrate and protein;(athlete) would need most energy/it releases most energy;for (contraction of) muscles;[max 2]
  - (iii)  $\frac{900}{129}$  (= 6.97(8));  $\times 100 = 697.6$  or 698 (cm<sup>3</sup>); [2]
  - (iv) pregnant women;
    extra calcium needed for bones of fetus;
    AVP e.g. osteoporosis in older women;
    [max 2]

[Total: 9]

4 (a)



complete circuit with no 'dead ends' or short circuits; on-off switch in main circuit using correct symbol; fuse in main circuit using correct symbol; [3]

- **(b) (i)** X marked in the heater branch, either side of heater; [1]
  - (ii) heated; bend; broken; [3]
  - (iii) in the air coming into the heater (owtte);senses/switches off when air temperature in room is too high/so it can check room temperature;[2]

[Total: 9]

Р	age 4	7	Mark Scheme	Syllabus	Paper
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5	(a)	(i)	bubbles (gently)/owtte;		[1]
		(ii)	zinc is above hydrogen in the reactivity series;		
		(,	zinc is below calcium in the reactivity series ;		
			zinc is above copper in the reactivity series;		[max 2]
	(b)	(i)	Increases;		[1]
		(ii)	any two from:		
			melts;		
			moves / floats ; bubbles ;		
			dissolves / disappears ;		
			and		
			more vigorous/flame/explodes;		[3]
					[Total: 7]
6	(a)	(i)	identical (with female adult aphid);		[1]
		(ii)	(the genetic information) will be identical;		[1]
		/:::\	(the grantic information is) different from the adult formals.		
		(iii)	(the genetic information is) different from the adult female; (the genetic information is) different from each other;		[2]
	(h)	(i)	phloem;		
	(10)	(1)	because it transports dissolved food substances around the plant;		[2]
		(ii)	one area of phloem correctly shaded;		[1]
	(c)	(i)	(carbon dioxide) + water $\rightarrow$ (sugar) + oxygen;		[1]
		(ii)	enough light;		
			(source of) carbon dioxide;		[2]
	(d)	(i)	the starch/blue-black colour is only found in the green areas;		[1]
		(ii)	because chlorophyll is needed for photosynthesis;		
		. ,	green areas contain chlorophyll;		<u>.</u>
			starch is produced in the green areas;		[max 1]
					[Total: 12]

Mark Scheme

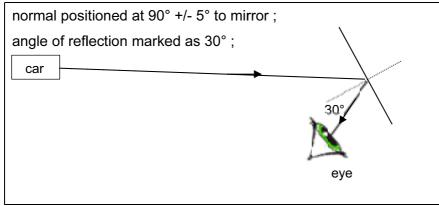
Syllabus

**Paper** 

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7 (a) (i)



[2]

(ii) image in a (plane) mirror is laterally inverted;

[1]

(b) microwaves;

[1]

(c) (i) labelled arrow to show amplitude correctly and clearly from x-axis to peak;

[1]

(ii) loudness/volume;

[1]

(d) distance travelled by sound =  $330 \times 2 = 660$  (m); distance from building =  $660 \times \frac{1}{2} = 330$  (m);

[2]

[2]

(e) by friction – (feet on carpet);discharge to Earth (when touches door handle) causes shock;

[Total: 10]

8 (a) neutrons/protons and protons/neutrons have the same mass, but electrons

have much smaller mass.;

have no charge.;

electrons/protons and protons/electrons have opposite charges but neutrons

[2]

(b) (i) (17)
number of electrons = number of protons / atom is has no overall charge; [1]

number of electrons = number of protons/atom is has no overall charge;

(ii) (18)
nucleon no. = proton no. + neutron no.; [1]

(iii) (non-metals are) on right-hand side of Periodic Table; [1]

(c) (i) shared electrons; [1]

(ii) bond between two non-metals; [1]

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	(d)	(i)	green;		[1]
		(ii)	7 to any less than 4 ;		[1]
	(e)	(i)	(a) salt ;		[1]
		(ii)	carbon dioxide ;		
			limewater; milky/cloudy;		[3]
					[Total: 10]
9	(a)	(i)	a chart/diagram showing the flow of energy; from organism to organism;		[2]
		(ii)	badger connected to all three organisms; arrows point in the correct direction;		[2]
	(b)		e an alternative source of food ; ve to a different habitat ;		[1]
					[Total: 5]